Application No. 09/632,812 Amendment dated September 10, 2003 Reply to Office Action of March 20, 2003

Client/Matter No. 09323.00001

Listing of Claims:

Claim 1 (previously presented) A process for regenerating a Type II strong base anion

exchange resin comprising:

passing an alkanolamine solution, whose effectiveness at removing H₂S and CO₂ from

gas streams has been decreased by the accumulation of heat stable salts, through a bed of Type II

strong base anion exchange resin until the active anion exchange sites of said Type II strong base

anion exchange resin are loaded with heat stable salt anions; and

contacting said loaded Type II resin with an amount of an alkali metal hydroxide and for

a time sufficient to obtain recovery of over 50% of the virgin capacity of the loaded Type II

resin.

Claim 2 (previously presented) The process according to claim 1 wherein said heat stable

salt anion is SCN.

Claim 3 (previously presented) The process according to claim 1 wherein said alkali

metal hydroxide is sodium hydroxide.

Claim 4 (previously presented) The process according to claim 3 wherein the amount of

sodium hydroxide is from about 1 to about 40 pounds of NaOH equivalent per cubic foot of

resin.

Claim 5 (previously presented) A process for regenerating a Type II strong base anion

exchange resin comprising:

passing an alkanolamine solution, whose effectiveness at removing H₂S and CO₂ from

gas streams has been decreased by the accumulation of heat stable salts, through a bed of Type II

strong base anion exchange resin until the active anion exchange sites of said Type II strong base

anion exchange resin are loaded with heat stable salt anions;

- 2 -

Application No. 09/632,812 Amendment dated September 10, 2003 Reply to Office Action of March 20, 2003

Client/Matter No. 09323.00001

contacting said loaded Type II resin with an amount of an alkali metal hydroxide and for

a time sufficient to obtain recovery of over 50% of the virgin capacity of the loaded Type II

resin; and

repeating the steps of loading said Type II resin with said anions and regenerating

repeatedly without substantial further reduction in active anion exchange sites.

Claim 6 (previously presented) The process according to claim 5 wherein said heat stable

salt anion is SCN⁻.

Claim 7 (previously presented) The process according to claim 5 wherein said alkali

metal hydroxide is sodium hydroxide.

Claim 8 (previously presented) A process consisting of:

loading a Type II strong base anion exchange resin with SCN;

washing said Type II anion resin with water;

regenerating said Type II anion exchange resin in a single step with a solution of sodium

hydroxide having a concentration of from about 1% to about 15% by weight of sodium

hydroxide at a temperature of from about 70°F, to about 120°F, in an amount of NaOH from

about 5 to about 35 pounds per cubic foot for from about 5 to about 120 minutes to remove heat

stable anions from said resin to obtain recovery of over 50% of the virgin capacity of the loaded

Type II resin; and

washing said Type II anion exchange resin with water.

Claim 9 (currently amended) A cyclic process for purifying an aqueous alkanolamine

solution containing alkali metal salts of anions which form heat stable salts with alkanolamines,

heat stable salts of such anions with alkanolamines, or both, comprising:

- 3 -

Application No. 09/632,812 Amendment dated September 10, 2003 Reply to Office Action of March 20, 2003 Client/Matter No. 09323.00001

(a) contacting the aqueous alkanolamine solution with a Type II strong base anion exchange resin to transfer at least some heat stable salt anions from the solution to

the resin;

(b) regenerating the resin by contacting the resin with an alkali metal hydroxide so

that the alkali metal hydroxide removes from the resin substantially all-the same

quantity of heat stable salt anions transferred to the resin in step (a); and

(c) repeating steps (a) and (b).

Claim 10 (previously presented) The process according to claim 9 wherein said alkali

metal hydroxide is sodium hydroxide.

Claim 11 (previously presented) The process according to claim 9 wherein said aqueous

alkanolamine solution is approximately 40% by weight alkanolamine.

Claim 12 (currently amended) A process for purifying an aqueous alkanolamine solution

containing alkali metal salts of anions which form heat stable salts with alkanolamines, heat

stable salts of such anions with alkanolamines, or both obtained from contacting the aqueous

alkanolamine solution with a hydrocarbon gas stream containing acid gasses, comprising:

(a) contacting the aqueous alkanolamine solution with a Type II strong base anion

exchange resin to transfer at least some heat stable salt anions from the solution to

the resin;

(b) recirculating the aqueous alkanolamine solution recovered from step (a) to contact

the hydrocarbon gas stream containing acid gasses;

(c) regenerating the resin by contacting the resin with an alkali metal hydroxide so

that the alkali metal hydroxide removes from the resin substantially all-the same

quantity of heat stable salt anions transferred to the resin in step (a); and

(d) repeating steps (a) - (c).

- 4 -

Application No. 09/632,812 Amendment dated September 10, 2003 Reply to Office Action of March 20, 2003 Client/Matter No. 09323.00001

Claim 13 (New) The method of claim 9, wherein the heat stable salts comprise thiocyanate.

Claim 14 (New) The method of claim 12, wherein the heat stable salts comprise thiocyanate.